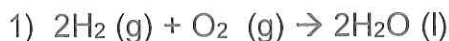
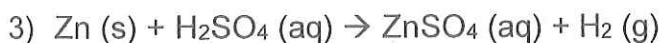
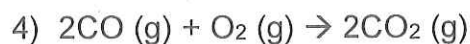
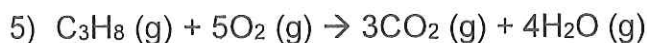
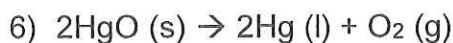
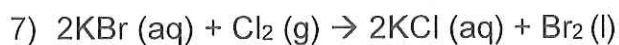
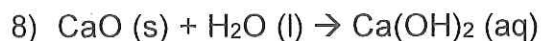
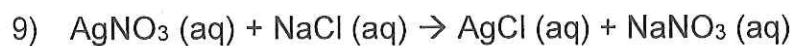
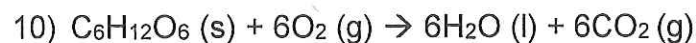
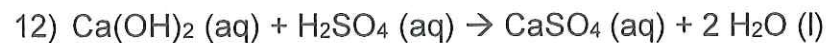
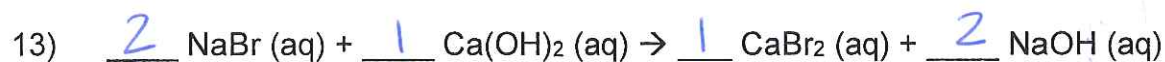


Name: Key Date: _____ Mods: _____*Unit 7: Classification of Chemical Reactions*

Classify the reactions below as synthesis, decomposition, hydrocarbon combustion, single replacement, or double replacement.

synthesisdecompositionsingle replacementsynthesiscombustiondecompositionsingle replacementsynthesisdouble replacementcombustiondecompositiondouble replacement

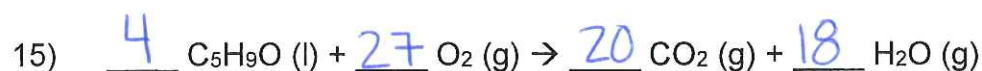
Directions: Balance the following reactions and indicate which of the six types of chemical reaction are being represented:



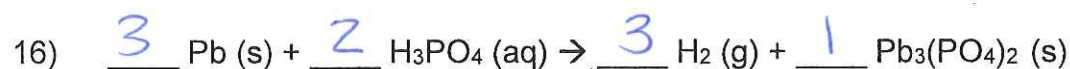
Type of reaction: double replacement



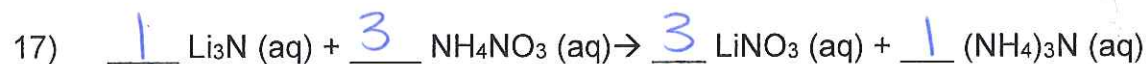
Type of reaction: synthesis



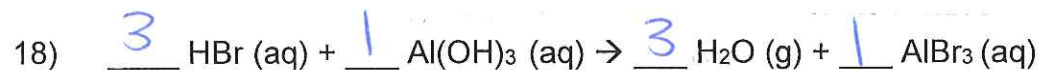
Type of reaction: combustion



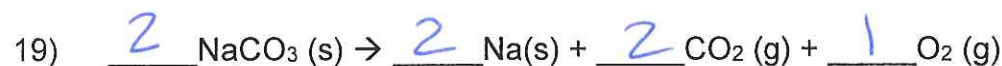
Type of reaction: single replacement



Type of reaction: double replacement



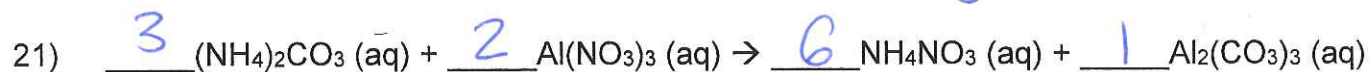
Type of reaction: double replacement



Type of reaction: decomposition



Type of reaction: synthesis



Type of reaction: double replacement